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ANSWER Answer the

following questions in
the space provided. 1.

Use the concept of
potential energy to
describe how a
covalent bond forms
between two atoms.

As the atoms involved
in the formation of a

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covalent bond

approach each other, the

6 1 Ionic

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SECTION 2 SHORT

ANSWER Answer the

following questions in

the space provided. 1.

Use the concept of

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potential energy to describe how a covalent bond forms between two atoms.

As the atoms involved in the formation of a covalent bond approach each other, the 6 Chemical Bonding Modern Chemistry Chapter 6 Vocab. 39 ...

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Borders. Standards.

SPS1. Students will investigate our current understanding of the atom. b. Compare and contrast ionic and covalent bonds in terms of electron movement. SPS2.

Students will explore

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the nature of matter,
its classification and
its system for naming
types of matter.

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Chapter Objectives.

Chapter 6 More on
Chemical Compounds

Study Guide Date

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proton nucleus level
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about the structure of
atoms. Label the
diagram of an atom.
Use these choices:
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Two or more atoms
held together by a
chemical bond is

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called what? answer
choices . A molecule.
A molecule. Tags:
Question 37 .

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chapter 6 section

3,4,5. ionic

compound. formula

unit. lattice energy.

polyatomic ion. A

compound that

consists of positive

and negative ions. the

lowest whole-number

ratio of ions in an

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The Structure of
Metals (pages
176–181) This section
discusses metallic
bonds and the
properties of metals. It

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also explains how the properties of an alloy are controlled.

Reading

Strategy (page 176)

Relating Cause and Effect As you read, complete the concept map to relate the structure of metals to their ...

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The Structure Of ...

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Answers By Karl May,
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answer answer the

following questions in

the space provided 1

a a chemical bond

between atoms

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results from the
attraction between the
valence electrons

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Chemical Bonding.

Chemical Bonding: •

Distinction between

elements and

compounds;

properties of each;

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same elements may form different compounds •

Reading chemical formulas: compound names, element names, number and ratio of atoms •

Chemical bonds between atoms involve electrons •

Ionic Bonds: result from electron transfer

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