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Professor Paul H. Emmett Chemistry Department
Portland State University P.O. Box 751
Portland, Oregon 97207 June 17, 1977 the fact
that nitrogen at -195°C is about 5%
imperfect at atmospheric pressure, the
correction being a linear function of the
pressure. Instead of a curve with kinks a
smooth S-shaped isotherm was obtained with
nitrogen at -195°C .

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of gases in ...~~

Paul Hugh Emmett was an American chemist best
known for his pioneering work in the field of

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catalysis and for his work on the Manhattan Project during World War II. He spearheaded the research to separate isotopes of uranium and to develop a corrosive uranium gas. Emmett also made significant contributions to BET Theory which explains the relationship between surface area and gas adsorption. He served on the faculty of Johns Hopkins University for 23 years throughout his scientific career.

~~Paul Hugh Emmett — Wikipedia~~

S. Brunauer, P. H. Emmett and E. Teller,
“Adsorption of Gases in Multimolecular

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Layers," Journal of the American Chemical Society, Vol. 60, No. 2, 1938, pp. 309-319.

~~S. Brunauer, P. H. Emmett and E. Teller,
"Adsorption of ...~~

The gas adsorption technique is based on the Brunauer-Emmett-Teller (BET) model included in the determination of the amount of gas that is adsorbed on the surface of the sample at low temperatures [13]. The specific surface area (m^2/g) defined by this method comprises the outward as well as the internal (open pores) surface area. However, the surface area of closed pores cannot be

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determined because of its inaccessibility for adsorbing gas molecules.

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Gases In~~

By making use of the evaporation-condensation properties of liquid mixtures, the Brunauer-Emmett-Teller theory of multimolecular adsorption is extended to mixtures of gases. No satisfactory experimental data are available at the present time with which to test the theory.

~~Theory of Multimolecular Adsorption from a
Mixture of ...~~

Emmett PH, Brunauer S, Love KS. The

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measurement of surface areas of soils and soil colloids by the use of low temperature van der waals adsorption isotherms Soil Science. 45: 57-65. 1: 1937: Emmett PH, Brunauer S. Accumulation of alkali promoters on surfaces of iron synthetic ammonia catalysts Journal of the American Chemical Society. 59: 310-315. 1: 1937

~~Paul Hugh Emmett — Publications~~

310 STEPHEN BRUNAUER, P. H. EMMETT AND EDWARD TELLER Vol. 60 DeB~er~'~ in several papers showed that various experimental adsorption isotherms could be fitted by equation (4a).

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However, they could not evaluate K_1 because in all cases, except one, the surface, and therefore v_m , was not known.

~~[CONTRIBUTION FROM THE BUREAU OF Adsorption
Gases in ...]~~

P. H. Emmett's 15 research works with 1,206 citations and 356 reads, including: THE APPLICATION OF POLANYI'S POTENTIAL THEORY TO THE VAN DER WAALS ADSORPTION OF GASES ON IRON SYNTHETIC AMMONIA ...

~~P. H. Emmett's research works~~

Brunauer-Emmett-Teller theory aims to explain

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the physical adsorption of gas molecules on a solid surface and serves as the basis for an important analysis technique for the measurement of the specific surface area of materials. The observations are very often referred to as physical adsorption or physisorption. In 1938, Stephen Brunauer, Paul Hugh Emmett, and Edward Teller published the first article about the BET theory in the Journal of the American Chemical Society. The BET theory ...

~~BET theory - Wikipedia~~

Stephen Brunauer, P. H. Emmett, and ; Edward

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Teller; Journal of the American Chemical Society 1938, ... Herbert S. Harned; Journal of the American Chemical Society 1938, ...

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Arwyn D. Evans, Matthew S. Cummings, Ryan Luebke, Martyn S. Brown, Silvia Favero, Martin P. Attfield, Flor Siperstein, David Fairen-Jimenez, Klaus Hellgardt, Russell Purves, David Law, Camille Petit. Screening Metal-Organic Frameworks for Dynamic CO/N₂ Separation Using Complementary Adsorption Measurement Techniques.

~~Adsorption of Gases in Multimolecular Layers
+ Journal of ...~~

Brunauer, S., Emmett, P.H. and Teller, E.
(1938) Adsorption of Gases in Multimolecular

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Layers. Journal of the American Chemical Society, 60, 309-319.

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Emmett P H & Teller E. Adsorption of gases in
multimolecular layers. Journal of the
American Chemical Society 60:309-19, 1938.
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Brunauer, S., Emmett, P.H., Teller, E.,
“Adsorption of gases in multimolecular
layers,” Journal of the American Chemical
Society, 60 (2). 309-319. 1938.

~~Brunauer, S., Emmett, P.H., Teller, E.,
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The Brunauer-Emmett-Teller (BET) equation
(Brunauer et al., 1938) overcomes the
limitation to a monolayer of sorbate and
currently is a commonly used model to

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describe the multilayer adsorption of gas molecules on a sorbent (Brunauer et al., 1938). For water adsorption, the BET equation can be written as

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